

MOLYBDENUM DISULFIDE ON REDUCED GRAPHENE OXIDE HYBRIDS WERE DEVELOPED VIA ONE POT HYDROTHERMAL ROUTE AS A CATALYST FOR HYDROGEN EVOLUTION REACTION

P.M. Premadasa¹, S.A. Panamldeniya¹, Y.K. Abeykoon¹, H.M.M. Munasinghe² and N. Gunawardhana^{1, 2*}

¹Postgraduate Institute of Science, University of Peradeniya, Peradeniya, Sri Lanka

²Sri Lanka Technological Campus, Padukka, Sri Lanka

*nandag@sltc.ac.lk

Green hydrogen production from the cathodic reaction of water splitting (hydrogen evolution reaction) is one of the most significant discoveries in recent years. In this regard, molybdenum disulfide (MoS₂) is used as an alternative catalyst for platinum due to its high availability and lower binding energy, which is close to zero. 1-T MoS₂, 3-R MoS₂, and 2-H MoS₂ are the major phases of MoS₂, and 2-H MoS₂ is the most stable form, but only the edge sites are activated, and the whole basal plane is almost inactive. Therefore, the activity of 2-H MoS₂ is based on the nature of the edges-sites. In this study, MoS₂-graphene oxide (MoS₂-GO) composites were prepared via a one-step hydrothermal method on the three distinct GO materials, which were prepared under different oxidation durations (24, 48, and 72 h) in the modified hummer's method. The MoS₂ with differently oxidized GO was characterized by SEM, PXRD, and FTIR to disclose the characteristics of the materials. In addition, the electrochemical performance of MoS₂-GO was investigated by using linear sweep voltammetry (LSV). The results showed that MoS₂ layers were successfully grown on the surface of GO, which is prepared at the highest oxidation duration (72 h). It has a higher number of shaped edges with uniform growth. According to the FTIR results of GO and GO-MoS₂, 72 h oxidized GO has more (COOH) carboxylic groups, and the growth of MoS₂ has been initiated through COOH groups. The XRD spectrum of the MoS₂/RGO revealed that the MoS₂ prepared on 72 h oxidized GO nanosheets (MoS₂/RGO-72) had the highest d spacing value due to the inserted GO into the MoS₂ layers. Moreover, MoS₂/RGO-72 was the best catalyst for the hydrogen evolution reaction (HER) because MoS₂/RGO-72 needs the lowest potential value to initiate the HER reaction (overpotential). Finally, the growth of MoS₂ on GO is possible to change by varying the oxidation duration of GO to have MoS₂/RGO material with a greater number of activated sharp edges.

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